

A MINI REVIEW ON APPLICATION OF TiO₂ NANOPARTICLE IN WOOL TECHNOLOGY

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ABSTRACT

In recent years attention has been towards the development of smart textiles with superior characteristics and enhanced durability. To cater the needs of high sustainable textiles, application of nanotechnology has been introduced in textile industry with promising achievement. Titanium oxide nanoparticle is one such material which could be applicable in textile industry, whose photo-catalytic activities have been widely investigated owing to its high stability and low toxicity. This paper enlists various applicative properties of titanium oxide nanoparticles and its implementation in the textile industry, especially to woollen fibres for advanced applications.

INTRODUCTION

The textile industry is no longer a mere supplier of fabrics, rather it is now evolving as a positive force that can help in the development of the society. Technology has made it possible to manufacture fabrics that have a potential to revolutionize our lives. Today the textile industry seeks innovations which can improve daily life, benefit the industry and health sector and adhere to eco-friendly norms. One such step towards development is development of fabrics that can clean themselves as well as purify water by using nothing but the sun as an energy source. This is possible by application of TiO₂ nanoparticle coating onto the fabric.

Titanium oxide (TiO₂) is a naturally occurring oxide of titanium, also known as Titania. The powdered ore of Titania is white in colour and is widely used as a pigment in paint and food industry after suitable purifications and modifications. The photo catalytic properties of TiO₂ was first discovered by Akira Fujishima in 1967 (Fujishima and Zhang, 2006). In recent years a lot of interest has been shown in the photo-catalytic potential of Titania due to its variant properties. Titania's photo-catalytic potential increases ten folds when used in nanoparticle form due to enhanced surface properties and the

several utilities linked to this property in TiO₂ has been illustrated below (Fig. 1). Photo-catalysis is a phenomenon where light acts as a catalyst to drive the reaction forward. In case of TiO₂ nanoparticle the photo catalytic activity is observed under UV spectrum (100nm-400nm) and the UV irradiation causes movement of positive holes and negative electrons, which have oxidising and reducing effects, respectively (Fujishima and Zhang, 2006; Nakajima, *et al.*, 2000). On one hand, holes oxidise water to hydroxyl (OH) ions, which are capable of manipulating configuration of dyes and other organic pollutants; while on the other hand electrons reduce oxygen molecules to superoxide radicles, which decompose harmful microbes, dirt, stains, etc (Ramasundaram, *et al.*, 2016; Pelaeza, *et al.*, 2012).

Thus, the application of TiO₂ nanoparticle onto a fabric, imparts properties such as self-cleaning, anti-aging, UV protection, antimicrobial, etc (Pekakis, *et al.*, 2006; Johnson, *et al.*, 2008; Munafò, *et al.*, 2014). In this article, we shall be focussing on the applications related to wool fibres. Wool, a widely used fibre in the textile industry, is acclaimed as a natural composite fibre with astounding manipulative chemical and physical properties, such as warmth, resiliency, fire-resistance etc. These properties are controlled

by both chemical composition as well as structural configuration of the fabric (Li, *et al.*, 2014; Quagliarini, *et al.*, 2012; Behzadnia, *et al.*, 2014) (Fig. 2).

However the wool fabric still has a number of limitations; low photo-stability, photo-yellowing, decomposition due to certain insects, microbial susceptibility, etc. (Mura, *et al.*, 2015; Montazer and Pakdel, 2010; Periolatto, *et al.*, 2013). These problems can be solved to a great extent by TiO₂ immobilization

(Yang, 2013; Kubacka, *et al.*, 2014; Gelover, *et al.*, 2006; Gómez-Ortiz, *et al.*, 2013; Lian, *et al.*, 2016).

RESULTS AND DISCUSSION

Application of TiO₂ nanoparticles

Titania is a special compound which finds a wide range of applications in various industrial sectors and Table 1 illustrates TiO₂ utilities discovered so far. The properties such as longer shelf-life, self-

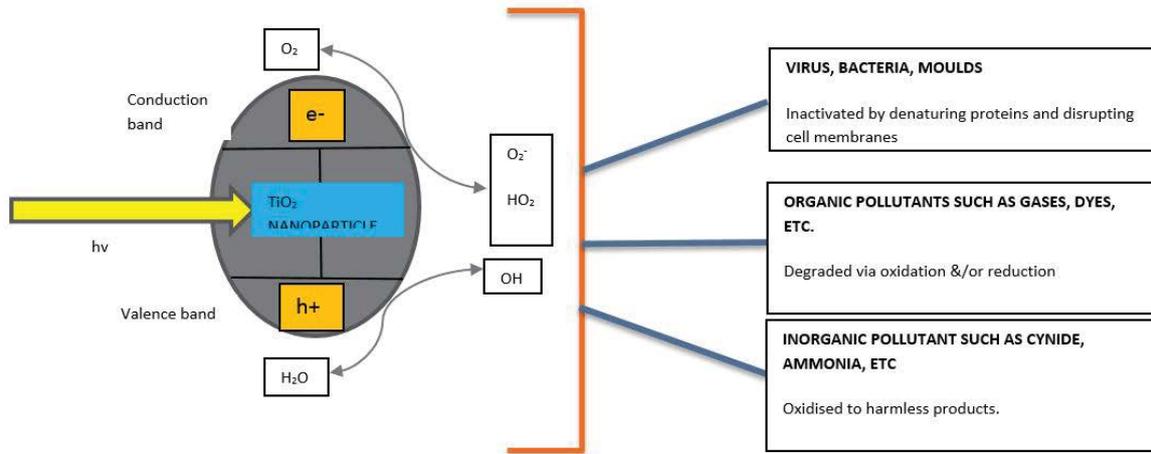


Fig. 1 Utility of the photo catalytic effect of TiO₂.

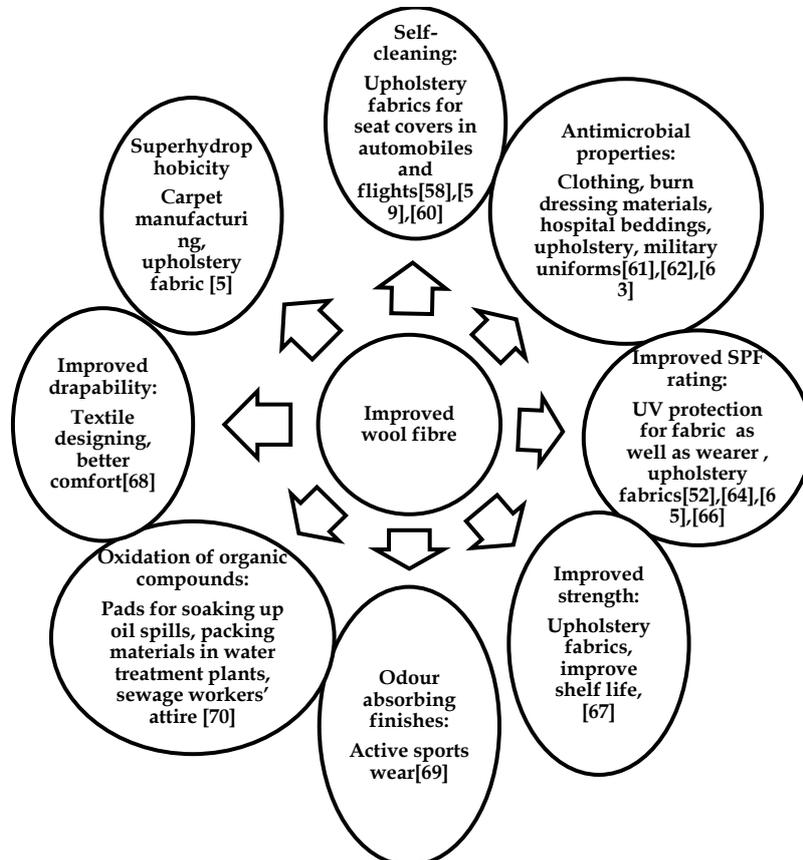


Fig. 2 Represents the various applications of immobilized TiO₂ on wool fiber.

Table 1. Various applications of TiO₂ nanoparticles

S. no.	Properties	Sector	Example	Ref
1	Self-Cleaning	Environment	Poly(vinyl dine fluoride) is used as a cross linker to immobilize TiO ₂ by melting at 160°C on the steel Mesh.SM helps to remove organic dyes from the flow of wastewater.	[2], [3], [4], [5]
			TiO ₂ /H ₂ O ₂ nanocomposite are used for the removal of the heavy metal from waste water.	[6]
			The bulk insertion of TiO ₂ in various materials (cements, ceramics etc.) related to building industry helps in degradation of organic and inorganic pollutant in gas phase.	[5], [7], [8], [9]
		Construction	Self-cleaning and anti-fogging glass is prepared due to the micro constructed composition of TiO ₂ at hydrophilic and oleophilic phases. Sometimes PEG and its composites are used a cross linker agent. Used in bulbs commonly.	[10], [11], [12], [17]
			TiO ₂ coating on the surface of lime stones protect heritage for several year.	[6], [14], [15]
			TiO ₂ addition in water based acrylic paint enhance its long lasting power and mechanical properties to cure the cracks etc.	[16], [17]
		Marine Hydrology	TiO ₂ coated Ti mesh works as oil separating device due to their superoleophobicity	[18]
		Food	Simultaneous Implication of UV radiation and TiO ₂ on sludge at high temperature helps in controlled removal of PAHs through the process of degradation.	[19]
Textile	Incorporation of TiO ₂ along with casein and other sensitizing agent in the different fabrics (e.g. cotton) protects the garments from organic stains.	[20], [21], [22]		
2	Anti UV property and anti-aging	Textile	Attachment of TiO ₂ nanoparticles with clothes to improve the anti-UV absorption and improve the life of material.	[23], [24]
3	Antimicrobial properties	Health	In rural areas, coliforms are removed from the water by using TiO ₂ coating on any substance, also could be used for anti-microbial agent towards pathogenic microbes.	[25], [26], [47]
		Construction	Ca (OH) ₂ and TiO ₂ mixture was coated on the limestone, prevents the growth of fungus.	[27]
		Food	In the presence of high pressure, TiO ₂ nanoparticle can be easily transported from polyvinyl-Chitosan biofilm or other packing material) to food stimuli (e.g olive oil) which helps to protect from biodegradation.	[28]
		Health	Coating of tior ₂ on the biomedical device basically prevents the transmission of various infectious diseases during the diagnosis period. Titanium based material could be used for space closure in bialveolar dental protrusion.	[29], [30]
4	Odour Removal	Health	Degrade the H ₂ S and other unpleasant odour bearing gas in NO ₂ ,SO ₂ and CO ₂	[31]

cleaning, UV protection and antimicrobial properties are much sought after to improve quality of life as well as create products that are sustainable and environment friendly (Visai, *et al.*, 2011; Chaitanya, *et al.*, 2017; Ao and Lee, 2005).

The main reasons for the wide acceptance of TiO₂ nanoparticles as a coating material its safety and non-toxicity to the fabric as well as to the skin, durability, ease of application, stability, no degradation on repetitive wash cycles, and the scope

for manipulations in its photo catalytic properties by modifying reaction conditions. (Tung and Daoud, 2009; Pakdel, *et al.*, 2013; Zhang, *et al.*, 2014; Li, *et al.*, 2010; Behzadnia, *et al.*, 2015; Euvananont, *et al.*, 2008) Moreover, it has been reported that it is possible to shift active region from UV range to visible light (Behzadnia, *et al.*, 2015). It is also applicable to a wide range of surfaces. In textile industry, it is used extensively due to its high compatibility with fabrics and low cost of production (Wang, *et al.*, 1997;

Table 2. Represents various techniques available for TiO₂ immobilization

Immobilization Techniques	Immobilizing agent	Conditions	Advantages	Disadvantages	Reference
Sol gel method	Succinyl anhydride	Room temperature	One step, easy and homogeneity at atomic level, High adhesion of TiO ₂ ,	Shrinkage, low durability, Deterioration of carrier gas	[32],[33],[34]
Hydrothermal method	Tetrabutyl titanate and ammonium chloride	Low temperature Presence of water in the system		More energy Requirement	[35],[20],[36]
Sono-synthesis		One step process in very optimum pressure and low temp(70°C-80°C)	Rapid, simple and inexpensive, Absence of toxic substances	No bond formation causes low adhesive nature	[37],[38],[39]
Self -assembly	poly (sodium 4-styrene-sulfonate) (PSS)	Electrostatic deposition	Less energy consumption, Easy and uniform distribution	Absence of covalent bond	[23]
Direct method		Lowering the pH at room temp		No bond formation	[40],
Grafting	Citric Acid		Most uniform deposition No degradation of carrier gas	May produce toxic waste	[41],[42]
Sputtering		Deposition of source material at a temperature lower than evaporation	Fast process	No bond formation only van der waal and mechanical interaction Less adhesive	[43]
Plasma treatment	Inert gas worked as a carrier gas	Occur at the room temp	Fast and one step process Strong adhesion	More energy consumption	[44],[45],[46]

Spasiano, *et al.*, 2015; Kim, *et al.*, 2016; Guan, 2005; Licciulli, *et al.*, 2011).

TiO₂ modifications and immobilisation onto woollen fabric

As mentioned earlier, wool is a photo-sensitive fibre and UV light, in particular, has a negative effect on the stability of the fibre. Immobilizing TiO₂ (coating the nanoparticle) on to the surface of the fabric helps solve these problems as well as improves the mechanical strength of the fabric (Ghoranneviss, *et al.*, 2011; Kan and Yuen, 2007; Kan and Yuen, 2006; Chi-wai, *et al.*, 2003; Chatterjee, *et al.*, 2016). Table 2 illustrates various modifications and conditions used for immobilizing TiO₂ onto woollen fabric (Ferrari, *et al.*, 2015; Carneiro, *et al.*, 2012; Salthammer and Fuhrmann, 2007; Banerjee, *et al.*, 2014; Karaca and Tasdemir, 2014).

Post application of TiO₂ nanoparticle coating properties such as improved strength, crease resistance, self-cleaning, UV protection, resistance against microbial decay, etc. have been achieved. Thus the newly improved fibre finds a greater deal of applications in diverse fields (Zhang, *et al.*, 2014; Xu, *et al.*, 2016; Kale, *et al.*, 2016; Liu, *et al.*, 2012).

CONCLUSION

Among the wide range of options available in nanoparticle applications, Titanium dioxide and its composites have gained most attention mainly due to their high chemical stability, ease of availability, low cost and non-toxicity. It significantly enhances the characteristics of wool fabric but does not alter any intrinsic property, mainly draping ability and resilience. Wool is a multifunctional fibre which comes in a range of diameters and fabrication techniques that makes it a widely sought after fibre for clothing manufacture, household fabrics as well as technical textiles. The application of TiO₂ nano-coating not only elevates the existing properties of the fibres but also imparts certain beneficial properties such as self-cleaning and odour. Overall outcome of such a modification would be the development of smart textiles with superior performance and enhanced comfort.

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